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## CH/AC/OC/CAH 402

## First Semester M.Sc. Degree Examination, December 2018 CHEMISTRY/APPLIED CHEMISTRY/ORGANIC CHEMISTRY/ANALYTICAL CHEMISTRY

(CBCS: 2016-17 Syllabus) Organic Chemistry

Time: 3 Hours Max. Marks: 70

**Note**: i) Answer **all** sub-questions from Part – **A** and **any four** questions from Part – **B**.

ii) Figures to the right indicate marks.

PART - A

1. Answer all the following sub-questions:

 $(2 \times 9 = 18)$ 

- a) How do you distinguish aromatic and non-aromatic compounds by chemical method?
- b) Which of the following pair is weak acid? Give reasons.

- c) What are cryptands? How are they differentiate from crown ethers?
- d) Which of the following alkyl halides would be most likely to give a rearranged product under  $S_N 1$  conditions? Give explanation.

$$Br$$
  $Br$   $Br$   $Br$   $Br$   $Br$ 

e) Predict the product(s) in the following reaction:

$$\begin{array}{c|c}
OMe \\
\hline
ONANH_2 \\
NH_3
\end{array}$$
?

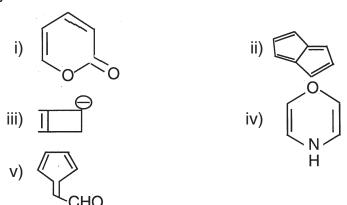


- f) Nitration of benzaldihyde and deuterated benzoldehyde occurs at the same rate. What does it indicate?
- g) Draw the most stable conformational structures of 2-methylcyclohexanone. Assign R/S configuration to their chiral centres.
- h) How many stereoisomeric products are obtained from the reaction of 1-methylcyclopentene with bromine? What is their relationship?
- i) Write the structures of
  - a) Z-3,6-diethyl-2,5-octadien-1-ol
  - b) Z-3-Phenyl-hex-3-en-(S)2-ol.

2. a) Explain the stability order for the following compounds

$$H_2C$$
 $X = CH_3$ 
 $= NH_2$ 
 $= OH$ 

b) Classify the following into aromatic, non-aromatic or anti aromatic with justifications.



- c) Discuss the inductive and resonance effects on the basic strengths of aryl amines. (4+5+4)
- 3. a) Calculate the energy associated with butodiene system.
  - b) Differentiate aromatic and non-aromatic compounds using spectroscopic methods.
  - c) Write a note on fluxional molecules.

(5+5+3)

- 4. a) Write a note on S<sub>N</sub>i reaction.
  - b) Explain the generation and reactivities of nitrenes.
  - c) Discuss the neighbouring group participation in nucleophilic substitution reactions. (4+5+4)



- a) Explain how the following techniques are useful in identifying reaction mechanism:
  - i) Isotopic labelling
  - ii) Trapping of intermediates.
  - b) Arrange the following free radicals in the order of increasing stability with justification.

c) Rank the following in order of decreasing rate of solvolysis with aqueous ethanol (fastest to slowest):

- 6. a) Reaction of (S)-2-methylcyclohexanone with NaBH4 in methanol will give how many products? Draw their conformational structures and indicate which is major.
  - b) Explain the optical activity in biphenyls and spiro-compounds.
  - c) Discuss the methods for determining the configuration of geometrical isomers. (3+5+5)
- 7. a) What is assymetric synthesis? Explain with examples.
  - b) Find and explain the relationship between Ha & Hb; Hb & Hc; and Hc and Ha in the following molecule.

$$\begin{array}{c} \text{CHO} \\ \text{HO} & & \text{H}_{\text{a}} \\ \text{H}_{\text{b}} & & \text{CH}_{\text{3}} \end{array}$$

c) Identify and write the structures of A and B in the following reactions.

Comment on the stereochemical outcome of the products. (3+6+4)

OH + (R) – 1 – Phenylethanol 
$$\longrightarrow$$
 A  $\xrightarrow{\text{PhMgBr}}$  B.